

Ya-Qian Cheng,\* Mao-Lin Hu,  
Sai-Rong Fan and Wei ZhangDepartment of Chemistry and Material Science,  
Wenzhou Normal College, Wenzhou 325027,  
People's Republic of ChinaCorrespondence e-mail:  
hu403cn@yahoo.com.cn

## Key indicators

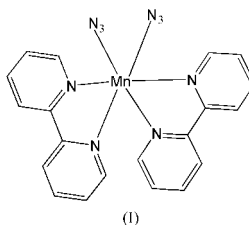
Single-crystal X-ray study  
 $T = 273\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$   
 $R$  factor = 0.059  
 $wR$  factor = 0.138  
Data-to-parameter ratio = 12.7For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

## Diazidobis(2,2'-bipyridine)manganese(II)

In the title compound,  $[\text{Mn}(\text{N}_3)_2(\text{C}_{10}\text{H}_8\text{N}_2)_2]$ , the Mn atom, on a twofold rotation axis, is chelated by the two heterocyclic ligands, with the two azide groups in *cis* positions in the distorted octahedral coordination.

## Comment

Considerable efforts have recently been concentrated on the synthesis and properties of complexes bridged by the azide anion, due to their intriguing structural diversity, and potential functions as models for metalloenzymes and ferromagnetic or antiferromagnetic interactions (Aebersold *et al.*, 1998; Baffert *et al.*, 2001). The azide anion can bind to metal ions in versatile modes: as a terminal ligand *via* one nitrogen donor, as a bridge in the  $\mu_{1,1}$  manner *via* one nitrogen donor, and in the  $\mu_{1,3}$  manner *via* both of the terminal *N*-donor atoms (Sheppard *et al.*, 1996). In our previous work, we have reported two monomeric complexes (Cheng & Hu, 2003; Cheng, Liu *et al.*, 2002) and a dimeric complex (Cheng, Hu *et al.*, 2002). Extending this research, we report here another new monomeric complex, namely diazidobis(2,2'-bipyridine)-manganese(II), (I).



In (I), two 2,2'-bipyridine (bpy) molecules, one manganese(II) cation and two azide anions constitute a monomeric unit, with Mn lying on a twofold rotation axis; the metal atom adopts an  $\text{MnN}_6$  octahedral geometry defined by four N atoms from two bpy ligands and two N atoms from two azide anions that occupy *cis* positions (Fig. 1). The basal plane  $[\text{N}2, \text{N}2^i, \text{N}3$  and  $\text{N}3^i$ ; symmetry code: (i)  $-x, y, \frac{z}{2} - z$ ] consists of two azide N atoms and two bpy N atoms, with bond lengths in the range 2.148 (3)–2.274 (3) Å. The apical positions are occupied by the other two bpy N atoms, with equal bond distances of 2.322 (3) Å and an angle of  $160.42(15)^\circ$ . Thus, the six-coordinate geometry of the manganese center can best be described as distorted octahedral, and is not significantly different from that found in the 2,2-bipyridine analog, which crystallizes as a monohydrate (Urtiaga *et al.*, 1995).

## Experimental

A mixture of manganese sulfate (1 mmol, 0.22 g), 2,2'-bipyridine (2 mmol, 0.31 g) and sodium azide (2 mmol, 0.13 g) in 50 ml of a

Received 12 January 2004

Accepted 16 January 2004

Online 23 January 2004

water and ethanol mixture (1:2 v:v) was stirred for several hours at room temperature. The mixture was filtered and red crystals separated from the filtrate after two weeks.

#### Crystal data

[Mn(N<sub>3</sub>)<sub>2</sub>(C<sub>10</sub>H<sub>8</sub>N<sub>2</sub>)<sub>2</sub>]

$M_r = 451.37$

Orthorhombic, *Pbcn*

$a = 14.5978$  (10) Å

$b = 8.1794$  (6) Å

$c = 16.9523$  (12) Å

$V = 2024.1$  (2) Å<sup>3</sup>

$Z = 4$

$D_x = 1.481$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

Cell parameters from 324 reflections

$\theta = 2.5$ – $22.0^\circ$

$\mu = 0.68$  mm<sup>-1</sup>

$T = 273$  (2) K

Prism, red

$0.42 \times 0.29 \times 0.28$  mm

#### Data collection

Bruker SMART CCD area-detector diffractometer

$\varphi$  and  $\omega$  scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2000)

$T_{\min} = 0.762$ ,  $T_{\max} = 0.832$

13540 measured reflections

1787 independent reflections

1751 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.025$

$\theta_{\text{max}} = 25.1^\circ$

$h = -17 \rightarrow 17$

$k = -9 \rightarrow 9$

$l = -20 \rightarrow 20$

#### Refinement

Refinement on  $F^2$

$R[F^2 > 2\sigma(F^2)] = 0.059$

$wR(F^2) = 0.138$

$S = 1.34$

1787 reflections

141 parameters

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0485P)^2 + 2.3037P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} < 0.001$

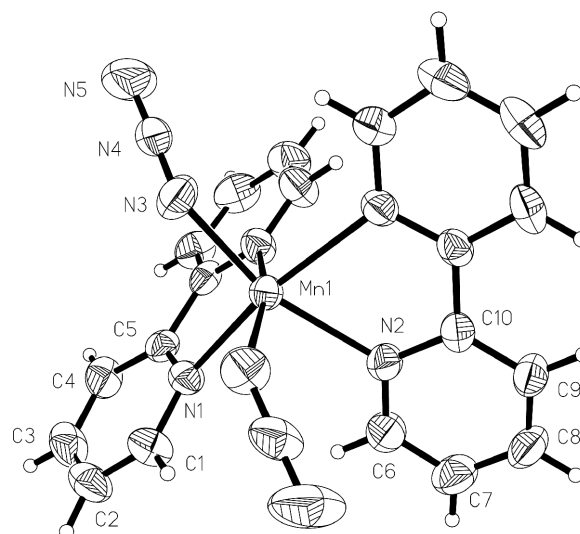
$\Delta\rho_{\text{max}} = 0.38$  e Å<sup>-3</sup>

$\Delta\rho_{\text{min}} = -0.29$  e Å<sup>-3</sup>

H atoms were positioned geometrically and allowed to ride on their parent atoms at a distance of 0.93 Å, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: *SMART* (Bruker, 2000); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 2000); program(s) used to solve structure: *SHELXTL* (Bruker, 2000); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

We acknowledge financial support by the Zhejiang Provincial Natural Science Foundation of China (No. 202137).



**Figure 1**

The molecular structure of (I), with the atom numbering, showing displacement ellipsoids at the 50% probability level.

#### References

- Aebbersold, M. A., Gillon, B., Plantevin, O., Pardi, L., Kahn, O., Bergerat, P., Seggern, I. V., Tuzcek, F., Ohrstrom, L., Grand, A. & Lelievre-berna, E. (1998). *J. Am. Chem. Soc.* **120**, 5238–5245.
- Baffert, C., Chen, H.-Y., Crabtree, R. H., Brudvig, G. W. & Collomb, M. N. (2001). *J. Electroanal. Chem.* **506**, 99–105.
- Bruker (2000). *SMART, SAINT, SADABS and SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Cheng, Y. Q. & Hu, M. L. (2003). *Z. Kristallogr. New Cryst. Struct.* **218**, 95–96.
- Cheng, Y. Q., Hu, M. L., Wang, S. & Ye, M. D. (2002). *Acta Cryst.* **C58**, m12–m13.
- Cheng, Y. Q., Liu, A. L., Hu, M. L. & Ng, S. W. (2002). *Acta Cryst.* **E58**, m545–m547.
- Sheppard, C. L., Tandon, S. S., Thompson, L. K., Bridson, J. N., Miller, D. O., Handa, M. & Lloret, F. (1996). *Inorg. Chim. Acta*, **250**, 227–239.
- Urriaga, M. K., Arriortua, M. I., De Muro, I. G. & Cortes, R. (1995). *Acta Cryst.* **C51**, 62–65.